

PART 1 - CHEMISTRY

Topic : Structure of atom

SECTION-1 : (Maximum Marks : 4)

- This section contains **ONE** question.
- Each question has **FOUR** options (A), (B), (C) and (D). **ONE OR MORE THAN ONE** of these four option(s) is(are) correct answer(s).
- For each question, choose the option(s) corresponding to (all) the correct answer(s).
- Answer to each question will be evaluated according to the following marking scheme :

Full Marks	: +4	ONLY if (all) the correct option(s) is(are) chosen;
Partial Marks	: +3	If all the four options are correct but ONLY three options are chosen;
Partial Marks	: +2	If three or more options are correct but ONLY two options are chosen, both of which are correct;
Partial Marks	: +1	If two or more options are correct but ONLY one option is chosen and it is a correct option;
Zero Marks	: 0	If none of the options is chosen (i.e. the question is unanswered);
Negative Marks	: -2	In all other cases.
- For example, in a question, if **(A), (B) and (D)** are the ONLY three options corresponding to correct answers, then choosing ONLY (A), (B) and (D) will get +4 marks; choosing ONLY (A) and (B) will get +2 marks; choosing ONLY (A) and (D) will get +2 marks; choosing ONLY (B) and (D) will get +2 marks; choosing ONLY (A) will get +1 mark; choosing ONLY (B) will get +1 mark; choosing ONLY (D) will get +1 mark; choosing no option (i.e. the question is unanswered) will get 0 marks; and choosing any other combination of options will get -2 marks.

1. The 2s and the 2p orbital energies of hydrogen atom are $E_{2s}(H)$ and $E_{2p}(H)$, respectively. The 2s and the 2p orbital energies of lithium atom are $E_{2s}(Li)$ and $E_{2p}(Li)$, respectively. The correct option(s) about the orbital energies is(are)
- [JEE Advanced 2026 Paper 1]
- (A) $E_{2s}(Li) < E_{2p}(Li)$ (B) $E_{2s}(H) = E_{2p}(H)$ (C) $E_{2p}(H) < E_{2s}(Li)$ (D) $E_{2s}(H) > E_{2s}(Li)$

Scan or tap for the
full solution



Space for Rough Work

**JEE ADVANCED STRUCTURE OF ATOM — ORBITAL
ENERGIES IN H AND MULTI-ELECTRON ATOMS — PYQS**

Test Pattern : JEE (Advanced)

ANSWER KEY**SECTION-1**

Ques.	1																
Ans.	A, B, D																