

Max. Marks: 4

Questions: 1

Q.1 Which of the following statements are correct?

- A. The process of adding an electron to a neutral gaseous atom is always exothermic.
- B. The process of removing an electron from an isolated gaseous atom is always endothermic.
- C. The 1st ionization energy of boron is less than that of beryllium.
- D. The electronegativity of C is 2.5 in CH_4 and CCl_4
- E. Li is the most electropositive among elements of group I.

(1) A, C and D Only (2) B and C Only
(3) B and D Only (4) B, C and E Only

Scan or tap for the full solution 

[illegible]