

JEE MAIN SOME BASIC CONCEPTS OF CHEMISTRY — EQUIVALENT
WEIGHT AND NORMALITY — PYQS

Max. Marks: 4

Questions: 1

Section B (Numerical) — 1 question

- Q.1** x mg of pure HCl was used to make an aqueous solution. 25.0 mL of 0.1 M Ba(OH)₂ solution is used when the HCl solution was titrated against it. The numerical value of x is _____ $\times 10^{-1}$. (Nearest integer)
Given : Molar mass of HCl and Ba(OH)₂ are 36.5 and 171.0 g mol⁻¹ respectively.

[JEE Main 2026 (Online) 23rd January Morning Shift]

Scan or tap for the
full solution

ANSWER KEY

Ques.	1							
Ans.	1825							