

NEET THERMODYNAMICS (PHYSICS) — SECOND LAW OF
THERMODYNAMICS — PYQS

Max. Marks: 4

Questions: 1

Section A (MCQ) — 1 question

Q.1 When 1 kg of ice at 0°C melts to water at 0°C , the resulting change in its entropy, taking latent heat of ice to be 80 cal/g , is [AIPMT 2011 (Prelims)]

- (1) 273 cal/K (2) $8 \times 10^4\text{ cal/K}$
(3) 80 cal/K (4) 293 cal/K

Scan or tap for the
full solution

ANSWER KEY

Ques.	1													
Ans.	4													